



0960-894X(95)00377-0

SYNTHESIS OF L-FUCOPYRANOSYL-4-THIODISACCHARIDES FROM LEVOGLUCOSENONE AND THEIR INHIBITORY ACTIVITY ON α -L-FUCOSIDASE.

Zbigniew J. Witczak*, Jianmin Sun and Raphael Mielguj

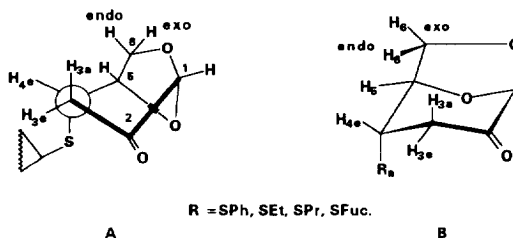
School of Pharmacy, Department of Pharmaceutical Sciences, University of Connecticut,
372 Fairfield Road, U-92, Storrs, CT 06269-2092, U.S.A.

Abstract: A new method of stereoselective synthesis of α -(1-4) linked thiodisaccharides: 4-S-(α -L-fucopyranosyl)-4-thio-3-deoxy- α -D-glucopyranose **6** and 2-acetamido-2,3-dideoxy-4-thio-(α -L-fucopyranosyl)-4-thio- α -D-glucopyranose **7**, by Michael addition of 1-thio- α -L-fucopyranoside to levoglucosenone is described. The inhibition study of **6** and **7**, against α -L-fucosidase (from bovine kidney) indicates the importance of 3-deoxy position of glucosamine moiety for total inhibitory effect $K_i=4.0\text{mM}$ and $K_i=2.4\text{mM}$ respectively.

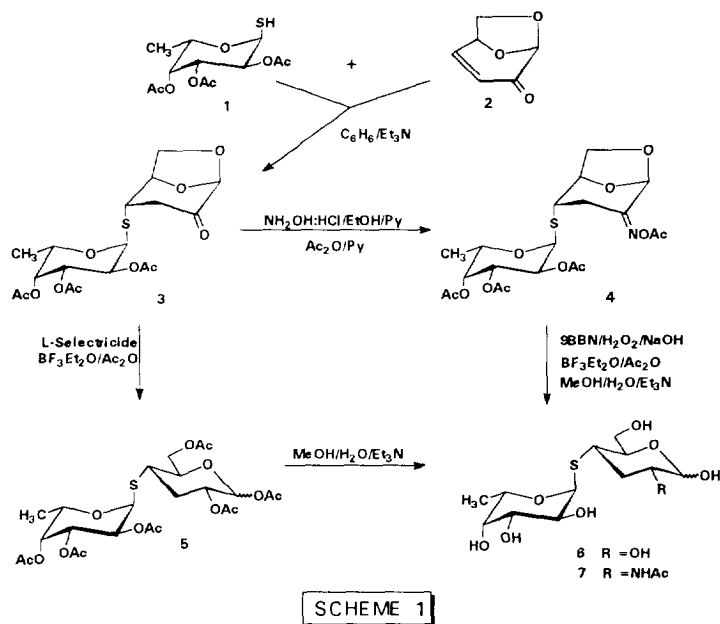
As part of our continuous interest in thiosugars^{1,2} as enzyme inhibitors³ and components of sugar antibiotics,⁴ we turned our attention to a new method of synthesis of α - and β -(1-4)-linked thiosugars³ containing biologically important sugar moieties such as galactose, glucose, mannose and L-fucose. Conventional methods are generally multistep and low overall yield approaches. The biological importance of the α -L-fucopyranosyl moiety is well defined in the literature.⁵ Specific inhibitors of α -L-fucosidase are potential anticancer agents especially against metastatic tumors.⁶ Recently, Hashimoto's group⁷ reported the synthesis and specific inhibitory activity of a few sulfur linked thiodisaccharides with 1-6, 1-4 and 1-3 linkages containing the α -L-fucopyranosyl moiety.

That report prompted us to test the inhibitory activity of earlier and presently synthesized (1-4)-S-thiodisaccharides⁸ in our laboratory. Our new approach to (1-4)-S-thiodisaccharides is based on the stereoselective Michael addition of 1-thio-sugars to the α,β - conjugated system of a convenient new chiral synthon levoglucosenone.⁹

The Michael addition of thiols to levoglucosenone was previously reported.¹⁰⁻¹¹ The shielding effect of the 1,6-anhydro bridge in levoglucosenone effectively prevents the formation of the 4-equatorial (*e*) product, and only the 4-axial (*a*) product was routinely obtained as a single addition product. This important fact



prompted us to explore the synthetic utility of levoglucosenone for the stereoselective introduction of a sulphur bridge connecting two sugar rings at C-(1-4). Indeed, the ^1H -NMR spin-spin coupling constants of the addition products indicate the C-4 as an axial position. The Newman projection (see A) of the adducts fully explains the geometry of the molecules. The most direct way to prove the correct stereochemistry of the 1,4-adduct is to measure the coupling constants, $J_{3\text{ax},4}$ and $J_{3\text{e},4}$ (see B), which range from 5.0-8.0 Hz and 1.0-1.5 Hz respectively. Lack of coupling between H-4 and H-5 indicates that the pyranose rings of the adduct is in the $^4\text{C}_4$ conformation, slightly distorted due to the presence of a carbonyl function at C-2 with an axial substituent at the 4-position. Due to long range couplings of 1-2 Hz, the complex multiplicity of the signals for H-3e is consistent with "W" planar arrangement of H-1, H-3, as well as H-5. This particular rule is highly predictable and has been observed by other authors¹² during the course of the Michael conjugate addition, as well as during



the base-catalyzed oligomerization of levoglucosenone.¹³ The addition reaction of thiol **1**¹⁴ was performed in polar solvent systems, preferably benzene or chloroform (CHCl_3), in the presence of a catalytic amount of triethylamine to produce **3**¹⁵ in 91% yields. The carbonyl function of **3** was stereoselectively reduced with L-Selectride¹⁶ followed by opening of the 1,6-anhydro ring through acid catalyzed acetolysis proceeding with the formation of exclusively D-ribo isomer **5**¹⁷ in 85% yields. The final deprotection by O-deacetylation ($\text{MeOH}/\text{Et}_3\text{N}/\text{H}_2\text{O}$) produced a mixture of anomers **6**¹⁸ in a 62% overall yield. The oximation of **3** with hydroxylamine hydrochloride in an ethanol/pyridine solution, followed by a conventional acetylation, produced acetoxime **4**¹⁹ in a 68% yield. The stereoselective reduction of acetoxime **4** with borane followed by acetylation with acetic anhydride in pyridine and deprotection by deacetylation, produced the *gluco*- isomer **7**²⁰ in an overall 79% yield. Only traces of the *manno*- isomer were detected by ^1H NMR.

This new approach constitutes a highly efficient and stereoselective methodology for the introduction of

the 1-4-thio bridge between two sugar units. Further efforts to extend the methodology to the generalization of various thio-sugar units by coupling with levoglucosenone are currently under extensive investigation in our laboratory. The replacement of the glucosidic linkage by a sulfur bridge will create a longer C-S bond a favorable attribute in a glycosidase inhibitor that mimics the transition state, which should have a "stretched" glycosidic C-O bond.²¹

The activity of **6** and **7** as inhibitors of the α -L-fucosidase from bovine kidney (EC3.2.1.51) are reported in Fig. 1 (Lineweaver-Burk plots) and in the comparison of K_i with the literature data^{5c,5d,5e,7} in Table 1. (K_i values [in mM] and the type of inhibitions).

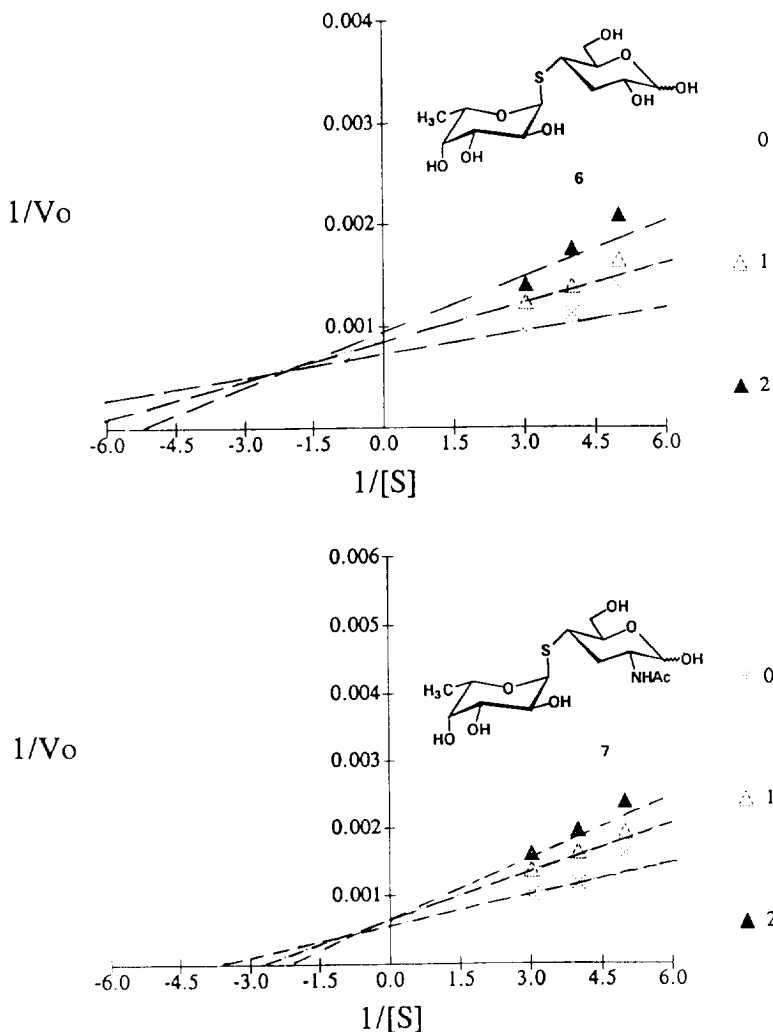


Figure 1. Lineweaver-Burk plots for the inhibition of α -L-fucosidase by compounds **6** and **7** at concentration 0 mM, 1 mM, and 2 mM of the inhibitor.

The preliminary enzyme inhibition assays of synthesized dithiasaccharides were determined by double-reciprocal plots ($1/V$ vs. $1/S$) for the hydrolysis of standard p-nitrophenyl- α -L-fucopyranoside against α -L-fucosidase (bovine kidney). The observed inhibitory activity for compounds **6** and **7**, which are 3-deoxy analogs of glucose and glucosamine as compared to the data reported by Hashimoto *et al.*⁷ suggests the importance of the above position particularly in the glucosamine moiety of compound **7**. Comparative inhibition studies of various analogs of **6** and **7** are under investigation in our laboratory.

Table 1. Inhibitory activities of α -(1-4)-thiodisaccharides **6** and **7** against α -L-fucosidase.^a

Compound	K_i values (mM)	Inhibition Type	Enzyme Source
6	4.0	mixed	
7	2.4	mixed	
Allyl 2-acetamido-2-deoxy-4-S-(α -L-fucopyranosyl)-4-thio- β -D-glucopyranoside ⁷	4.9	mixed	
L-fucose ^{5c}	0.30	competitive	
5-thio- α -L-fucose ^{5d}	0.042	competitive	
	0.084	competitive	bovine epididymis
1,5-dideoxy-1,5-imino-L-talitol ^{5c}	0.007	competitive	
	0.011	competitive	bovine epididymis
α -L-homofuconojirimycin ^{5f}	5.8 (nM)	competitive	bovine epididymis
deoxyfuconojirimycin ^{5f}	6.2 (nM)	competitive	bovine epididymis

^a α -L-Fucosidase from bovine kidney (EC3.2.1.51) was purchased from Sigma Chemical Co. The enzyme assay was performed²² by essentially the same method as that of Evans *et al.*²³ The inhibition modes were determined by Lineweaver-Burk plots. The constant K_m was calculated with the program Enzyme Kinetics (Windows Chem. Software). The K_i values were calculated by plotting the apparent K_m values versus the inhibitor concentration.

ACKNOWLEDGMENT. Financial support from the University of Connecticut Research Foundation and State of Connecticut under the "Windows of Opportunity" grants program is gratefully acknowledged.

REFERENCES AND NOTES

1. Witczak, Z. J. *Adv. Carbohydr. Chem. Biochem.* **1986**, *44*, 91.
2. Witczak, Z. J. *Tetrahedron Lett.* **1986**, *27*, 155.
3. For review see: (a) Defaye, J.; Gelas, J. In *Studies in Natural Products Chemistry*; Atta-ur-Rahman, Ed.; Elsevier: Amsterdam, 1991; Vol 8E, pp. 315-357; For the synthesis of S-linked sugars see: (a) Hutson, D.M., *J. Chem. Soc. C.* **1967**, 442; (b) Blanc-Muesses, M.; Defaye, J.; Driguez, H. *Tetrahedron Lett.* **1976**, *47*, 4307; (c) Blanc-Muesser, M.; Vigne, L.; Driguez, H.; Lehmann, J.; Steck, J.; Urbahns, K. *Carbohydr. Res.* **1992**, *224*, 59; (d) Wand, L. X.; Sakari, N.; Kuzuhara, H. *J. Chem. Soc. Perkin Trans. 1*, **1990**, 1677; (e) Reed, L. A.; Goodman, L. *Carbohydr. Res.* **1981**, *94*, 91; (f) Andrews, J. S.; Johnston, B. D.; Pinto, B. M. *J. Am. Chem. Soc.* **1994**, *116*, 1569.

4. For thio-sugar antibiotics see: Golik, J.; Wong, H.; Vyas D.M.; Doyle, T.W. *Tetrahedron Lett.* **1990**, 30, 2497-2500; *Tetrahedron Lett.* **1991**, 32, 1851.
5. (a) Winchester, B.; Baker, C.; Baines, S.; Jacob, G. S.; Namgong, S. K.; Fleet, G. W. J. *Biochem J.* **1990**, 265, 277; (b) Cai, S.; Stroud, M. R.; Hakomori, S.; Toyokumi, T. *J. Org. Chem.* **1992**, 57, 6693; (c) Dumas, D. P.; Kajimoto, T.; Liu, K. K. C.; Wong, C. H.; Berkowitz, D. B.; Danishefsky, S. J. *Bioorg. Med. Chem. Lett.* **1992**, 2, 33; (d) Hashimoto, H.; Fujimori, T.; Yuasa, H. *J. Carbohydr. Chem.* **1990**, 9, 683; (e) Hashimoto, H.; Hayakawa, M. *Chem. Lett.* **1989**, 1881. (f) Andrews, D. M.; Bird, M. I.; Cunningham, M. M.; Ward, P. *Bioorg. Med. Chem. Lett.* **1993**, 3, 2533.
6. Muramatsu, T. *Glycobiology* **1993**, 3, 291.
7. Hashimoto, H.; Shimada, K.; Horito, S. *Tetrahedron Lett.* **1993**, 34, 4953; *Tetrahedron: Asymmetry* **1994**, 5, 2351.
8. Witczak, Z. J. In *Levogluconone and Levoglucosans Chemistry and Applications*; Witczak, Z. J. Ed.; ATL Press: Mount Prospect, 1994; pp. 7-8. Synthesis of another (1-4)-S-linked thiodisaccharides composing, of galactose, fucose and mannose, will be reported in due course.
9. (a) Shafizadeh, F.; Chin, P. P. P. *Carbohydr. Res.* **1977**, 58, 79; (b) For review see: Witczak, Z. J. In *Studies in Natural Products Chemistry*; Atta-ur-Rahman, Ed.; Elsevier: Amsterdam, 1994, Vol 14, pp. 268-282.
10. Shafizadeh, F.; Furneaux, R. H.; Stevenson, T. T. *Carbohydr. Res.* **1979**, 71, 169.
11. Essig, M. G. *Carbohydr. Res.* **1986**, 156, 225.
12. (a) Forsyth, A.C.; Paton, M.R.; Watt, I. *Tetrahedron Lett.* **1989**, 30, 993; (b) Forsyth, A. C.; Gould, R.O.; Paton, R.M.; Sadler, I.H.; Watt, I. *J. Chem. Soc. Perkin Trans. 1*, **1993**, 2737;
13. Shafizadeh, F.; Furneaux, R. H.; Pang, P.; Stevenson, T. T. *Carbohydr. Res.* **1982**, 100, 303.
14. Thio sugar **1** was prepared by the method of Matta, K. L.; Girotra, R. N.; Barlow, J. J. *Carbohydr. Res.* **1975**, 43, 101, with the modification at the final stage using method of Joseph, B.; Rollin, P. J. *Carbohydr. Chem.* **1993**, 12, 719. Syrup [α]²³ - 165° (c 1.00 CHCl₃). Lit. ⁷ [α]²³ - 162° (c 1.00 CHCl₃); For ¹³C NMR see ref. 7.
15. Compound **3**, 1,6-anhydro-3-deoxy-4-S-(2,3,4,-tri-O-acetyl- α -L-fucopyranosyl)-D-glycero-hexo-pyranos-2-ulose, syrup [α]²³ - 153° (c 1.00 CHCl₃); ¹H NMR (250MHz, CDCl₃): δ 5.66 (d, 1H, J_{1,2} = 5.8Hz, H-1'), 5.16 (s, 1H, H-1), 4.72 (m, 1H, H-5), 5.06 (dd, 1H, J_{2,3} = 11.2Hz, J_{3,4} = 3Hz, H-3'), 4.04 (dd, 1H, J_{5,6e} = 1.6Hz, J_{6a,6e} = 12Hz, H-6e), 4.23 (q, 1H, J_{5',6'} = 6Hz, H-5'), 3.45 (d, J_{3a,4} = 8Hz, H-4), 3.12 (dd, 1H, J_{3a,4} = 8Hz, H-3a), 2.58 (d, 1H, J_{3a,3e} = 16.4Hz, J_{3a,4} = 7.9Hz, J_{3e,4} = 0.8Hz, H-3e), 2.06, 2.10, 2.12, (3s, 9H, 3Ac); ¹³C NMR: (CDCl₃): δ 17.5 (CH₃-C-6'), 20.9, 20.8, 20.6 (CH₃CO-), 39.8 (C-3), 44.2 (C-4), 67.8 (C-6), 70.9 (C-5'), 74.8 (C-3'), 77.6 (C-5), 77.9 (C-4'), 83.3 (C-2'), 98.2 (C-1'), 101.3 (C-1), 170.6, 170.3, 170.1 (-COCH₃), 197.9 (C-2).
16. Okano, K.; Ebata, T.; Koseki, K.; Kawakami, H.; Matsumoto, K.; Matsushita, H.; *Chem. Pharm. Bull.* **1993**, 41, 861.
17. Compound **5**, acetyl, 3-deoxy-4-S-(2,3,4,-tri-O-acetyl- α -L-fucopyranosyl)- α , β -3,6-di-O-acetyl-D-glucopyranose, syrup [α]²³-165° (c 1.00 CHCl₃). ¹H NMR (250MHz, CDCl₃): δ 5.68 (d, 1H, J_{1,2} = 5.6Hz, H-1'), 5.09 (dd, 1H, J_{2,3} = 11.0Hz, J_{3,4} = 3.2Hz, H-3'), 4.68 (m, 1H, H-5), 4.48 (d, 1H, H-1), 4.38 (q 1H, J_{5',6'} = 5.8Hz, H-5'), 3.86 (dd, 1H, J_{5,6a} = 5Hz, J_{6a,6e} = 12Hz, H-6e), 3.78 (d, 1H, J_{3',4'} = 3Hz, H-4'), 3.62-3.42 (m, 3H, H-2, 5, 3'), 3.2 (dd, 1H, H-3a), 2.68 (m, 1H, H-3e), 2.01, 2.04, 2.06, 2.08,

- 2.10, 2.12, (6s, 18H, 6Ac), 1.18 (s, 1H, CH₃-C-6'); ¹³C NMR (CDCl₃): δ 16.1 (CH₃-C-6'), 20.4, 20.5, 20.6, 20.7, 20.8, 20.9, (-COCH₃), 39.6 (C-3), 46.2 (C-4), 61.8 (C-6), 66.8 (C-5'), 68.6 (C-3'), 72.2 (C-4'), 74.8 (C-5), 96.2 (C-1'), 101.1 (C-1), 169.8, 170.0, 170.1, 170.4, 170.8 (-COCH₃).
18. The anomeric mixture **6** of 3-deoxy-4-S-(α-L-fucopyranosyl)-α,β-D-glucopyranose was separated by flash column chromatography on silica gel by elution with AcOEt/Hexane 3:1 v/v. ¹H NMR 250MHz, CDCl₃): δ 5.42 (d, 1H, J_{1,2} = 5.2Hz, H-1'), 4.83 (q, 1H, J_{5,6} = 5.8Hz, H-5'), 4.75 (d, 1H, H-1 α-anomer), 4.48 (d, 1H, H-1, β-anomer), 4.42 (q, 1H, J_{5',6'} = 6Hz, H-5'), 4.18-4.04 (m, 2H, H-2', H-6a), 3.8 (dd, 1H, J_{5,6a} = 5Hz, J_{6a,6e} = 12.6Hz, H-6e), 3.72 (d, 1H, J_{3',4'} = 2.8Hz, H-4'), 3.68-3.48 (m, 3H, H-2,5,3'), 3.10 (dd, 1H, H-3a), 2.56 (d, J_{3a,3e} = 16Hz, J_{3e,4} = 7.9Hz, 1H, H-3e), 1.16 (s, 1H, CH₃-C-6'); ¹³C NMR (D₂O): (α-anomer): δ 15.9 (CH₃-C-6'), 46.4 (C-3), 60.4 (C-6), 66.8 (C-5'), 67.6 (C-2'), 69.2 (C-3'), 71.8 (C-4'), 73.6 (C-2), 74.6 (C-3), 74.9 (C-5), 99.1 (C-1'), 100.6 (C-1); (β-anomer): 102.3 (C-1).
19. Compound **4**, 1,6-anhydro-2-acetoximino-2,3,-dideoxy-4-S-(2,3,4-tri-O-acetyl-α-L-fucopyranosyl)-α,β-D-glycero-hexo-pyranose, syrup [α]_D²³-159° (c 1.00 CHCl₃). ¹H NMR (250MHz, CDCl₃): δ 5.59 (d, 1H, J_{1',2'} = 5.4Hz, H-1'), 5.16 (s, 1H, H-1), 5.1 (dd, 1H, J_{2',3'} = 11.0Hz, J_{3,4} = 3.2Hz, H-3'), 4.76 (m, 1H, H-5), 4.02 (dd, 1H, J_{5,6a} = 1.8Hz, J_{6a,6e} = 12.0Hz, H-6e), 4.48 (q, 1H, J_{5',6'} = 6Hz, H-1), 3.78 (d, 1H, J_{3',4'} = 2Hz, H-4'), 3.12 (dd, 1H, H-3a), 2.54 (d, 1H, J_{3a,3e} = 16.2Hz, J_{3a,4} = 7.8Hz, J_{3e,4} = 0.9Hz, H-3e), 2.04, 2.06, 2.08 (3s, 9H, 3Ac), 2.2 (s, 3H, -NOAc), 1.23 (d, 3H, CH₃-6'); ¹³C NMR: (CDCl₃): δ 16.1 (CH₃-C-6'), 20.6, 20.8, 20.9, (CH₃CO-), 22.6 (NHCOCH₃), 39.4 (C-3), 44.6 (C-4), 66.8 (C-5'), 67.8 (C-2'), 68.2 (C-6), 69.2 (C-3'), 72.5 (C-4'), 77.2 (C-5), 98.3 (C-1'), 101.6 (C-1), 187.3 (C-2).
20. Compound **7**, 2-acetamido-2-deoxy-4-S-(α-L-fucopyranosyl)-4-thio-α,β-D-glucopyranoside, syrup [α]_D²³ -201° (c 1.00 H₂O); ¹H NMR (250MHz, D₂O): δ 5.8 (d, 1H, J_{2,NH} = 8.9Hz, NH), 5.36 (d, 1H, J_{1',2'} = 5.2Hz, H-1'), 4.5 (d, 1H, J_{1,2} = 8.4Hz, H-1), 4.32 (q, 1H, J_{5',6'} = 6.0Hz, H-5'), 4.2-4.06 (m, 2H, H-6a, H-2'), 3.86 (dd, 1H, J_{5,6a} = 5Hz, J_{6a,6e} = 12Hz, H-6e), 3.78 (d, 1H, J_{3',4'} = 2.6Hz, H-4'), 3.71-3.52 (m, 3H, H-2,5,3'), 2.72 (t, 1H, J_{3,4} = J_{4,5} = 10Hz, H-4), 2.02 (s, 3H, AcO), 1.23 (d, 3H, CH₃-6'); ¹³C NMR (D₂O): δ 15.2 (CH₃-C-6'), 21.8 (COCH₃), 38.2 (C-3), 44.6 (C-4), 56.8 (C-2), 61.4 (C-6), 66.6 (C-5'), 68.2 (C-2'), 68.8 (C-3'), 71.8 (C-4'), 74.6 (C-5), 95.2 (C-1'), 100.1 (C-1), 173.8 (NHCOCH₃).
21. Liu, P.S. *J. Org. Chem.* **1987**, 52, 4717.
22. Inhibition assays were performed at 25 °C for 45 min in 20 mM citrate buffer (pH 5.5 for bovine kidney, 300 μL), which contained the following assay components: p-nitrophenyl-α-fucopyranoside (0.07-0.33 mM), the S-linked disaccharide **6**, **7** (each 0-6.7 mM.), α-L-fucosidase (0.8-1.1 units/mL from bovine kidney. Initial velocities (less than 10% substrate consumed) were measured by quenching the reaction after 45 min by the addition of 50 mL of glycine buffer (pH 10.00, 500 μL and determining the p-nitrophenolate ion concentration from its absorption at 400 nm (reference, same solution without enzyme). The data were expressed by Lineweaver-Burk plots and respective replots of the slope versus the inhibitor concentration and the intercept versus the inhibitor concentration indicate the modes of inhibition and give the K_i values.
23. Evans S. V.; Fellows, L. E.; Shing, T. K. M.; Fleet, G. W. J. *Phytochemistry* **1985**, 24, 1953.